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Chemists combine all of the relationships seen in Models 1 and 2 into one law—the Ideal Gas Law. It is one equation that describes gas behavior and the relationship among all four variables, P, V, T, and n. In the Ideal Gas Law the proportionality constant is represented by the letter R (rather than the generic k). 24.

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a. Identify each of the following variables: Controlled Current of the cell. Independent The time the cell is run. Dependent Mass of Ag collected. b. Describe the relationship between the independent and dependent variable in this experiment. Linear? Inverse? Exponential? Logarithmic? Justify your answer. The relationship is a direct linear ...

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Chem 116 POGIL Worksheet - Week 1 Gas Laws - Part 1 Why? Of the three principal states of matter (gas, liquid, solid), gases show behavior that is most easily connected to molecular motion. The observed behavior of gases, embodied in the empirical gas laws, leads to a series of equations that can be summarized by a single equation of state, called

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Chem 116 POGIL Worksheet - Week 1 - Solutions Gas Laws - Part 1 Key Questions 1. Using SI units of kilograms, meters, and seconds with these fundamental equations, determine the combination of units that define the following: i. The newton (N), the fundamental unit of force $F = ma = (kg)(m@s^{-2}) = kg@m@s^{-2} = N$ ii.

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